## Carbon–negative and Sustainable CO<sub>2</sub> Conversion to Value– added Chemicals using Microbial Electrosynthesis

### 신재생전기에너지와 미생물전기합성 기반 CO<sub>2</sub> 전환을 통한 탄소네가티브 고부가가치 화학물질 생산



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## **Bioelectrochemical Sysetem (BES)**

BES converts chemical energy of organic waste, biomass into electricity by **microbial fuel cell (MFC)** or produce hydrogen/chemical products through **microbial electrosynthesis cells (MEC or MES)** 



## **Electrochemically Active Bacteria**





### 전기화학적 활성 미생물



### **Extracellular Electron Transport**





#### Kracke et al. Frontiers in Microbiology 2015 5

## **Microbe-Electrode Hybrid System**



- Live cell continuously interacts with electrode to exchange respiratory electrons
- Applied potential influences gene expression and metabolic pathway

### Microbial fuel cell & Microbial electrosynthesis cell

#### Microbial fuel cell (MFC)



- Spontaneous oxidation/reduction
- Microbial catalyst on anode
- Production of electricity from biodegradable organic materials and waste

#### Microbial electrosynthesis cell (MEC)



- Non-spontaneous oxidation/reduction
- Microbial catalyst on cathode
- H<sub>2</sub>, CH<sub>4</sub>, VFA and platform chemical production by electrical energy
- MES can be used for upgrading of CO<sub>2</sub>

## **Renewable Energy vs. Electricity**







WIND



TIDAL

SOLAR



GEOTHERMAL







Electricity is key of future energy infra for production/storage /transportation/utilization



Notes: CSP = concentrating solar power; TWh=terawatt hour.

## **Demand of Fossil Oil for Refinery**



## **Carbon Capture & Utilization (CCU)**





DOI:10.2118/194190-PA

## **Carbon Neutral Process by CO<sub>2</sub> Valorization**

$\begin{array}{c} \hline CO_2 \end{array} \rightarrow \begin{array}{c} \hline Utilization \\ \hline $						
생성물	반응	필요한 전 자의 개수	<i>E°</i> (V vs. SHE)			
со	$\mathrm{CO_2}\ +\ 2\mathrm{H}^+\ +2\mathrm{e}^- \rightarrow \mathrm{CO}\ +\ \mathrm{H_2O}$		-0.53			
Formate	$2CO_2 + 2H^+ + 2e^- \rightarrow HCOOH$	2e-	-0.61			
Oxalate	$2\text{CO}_2 + 2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2\text{C}_2\text{O}_4$		-0.913			
Formaldehyde	$\mathrm{CO_2}\ +\ 4\mathrm{H^+}\ +4\mathrm{e^-} \rightarrow \mathrm{HCHO}\ +\ \mathrm{H_2O}$	4e⁻	-0.48			
Methanol	$\mathrm{CO}_2 \ + \ \mathrm{6H^+} \ + \mathrm{6e^-} \ \rightarrow \ \mathrm{CH_3OH} \ + \ \mathrm{H_2O} \ \qquad \mathrm{6e^-}$		-0.38			
Methan	$\mathrm{CO_2}\ +\ 8\mathrm{H^+}\ +8\mathrm{e^-}\ \rightarrow\ \mathrm{CH_4}\ +2\mathrm{H_2O}$	8e-	-0.24			
Ethylene	$2CO_2 + 12H^+ + 12e^- \rightarrow C_2H_4 + 4H_2O$	-0.349				
Ethanol	$2CO_2 + 12H^+ + 12e^- \rightarrow C_2H_5OH + 3H_2O$	12e	-0.329			
Ethane	$2CO_2 + 14H^+ + 14e^- \rightarrow C_2H_6 + 4H_2O$	14e-	-0.27			
Propanol	$3CO_2 + 18H^+ + 18e^- \rightarrow C_3H_7OH + 3H_2O$	18e-	-0.31			

#### • CO<sub>2</sub> is very stable and most oxidized form of carbon

#### • Electron/reducing power is required for conversion

#### Electrocatalytic CO<sub>2</sub>ER reaction pathways



	Advantage	Disadvantage	
Chemical Catalyst	Fast reaction rate, Study of reaction mechanism	High temp/pressure, Limited products, Catalyst poison, Limited lifecycle	
Biocatalyst	<u>Catalyst reproducible</u> , High value- added products, <u>Room Temp/pressure</u> , Improvement of performance by OMICS	Low reaction rate, Difficult to control the mechanism	

### Power to Product (P2P) for CO<sub>2</sub> Conversion

#### **Electricity-based single cell protein (SCP) by P2P and CCU**



Renewable hydrogen based sustainable protein production

Hydrogen electrolyzer is a major platform for Power to X





## Power to Product (P2P) for CO<sub>2</sub> Conversion

#### Three different SCP production strategies



- The PV-driven SCP produces more protein per hectare as compared to sugar beet SCP and conventional soybean agriculture
- The protein yields and amount of people that could be fed from 1 ha
- Assumption of an irradiance of 2,000 kWh·m<sup>-2</sup>·y<sup>-1</sup>

## **Microbial Electrosynthesis Cell (MEC)**





Young Eun Song and Jung Rae Kim. 2022. Chemical Engineering Journal. 427:131885

## Acetate Production from CO<sub>2</sub> in MEC





Young Eun Song and Jung Rae Kim. 2022. Chemical Engineering Journal. 427: 131885

## **Metabolic Engineering for Acetate Conversion**



ACS Publications

www.acs.org

Mutyala and Kim. 2023. ACS Omega. 8:29. 26231-26242



# CO<sub>2</sub> to H<sub>2</sub> with *Rhodobacter sphaeroides*



### R. sphaeroides

- Photoheterotrophic H<sub>2</sub>
  production
- No Photosystem II Light energy → ATP (PS1)



#### Photoautotrophic CO<sub>2</sub> conversion & H<sub>2</sub> production *R. sphaeroides*



- -0.9V vs. Ag/AgCl applied
  - under light (5000 lux)
- Consumption of CO<sub>2</sub> and

simultaneously H<sub>2</sub>

production

CO<sub>2</sub> 100%

Shuwei Li and Jung Rae Kim. 2021. Bioresource Technology. 320:124333

# CO<sub>2</sub> to H<sub>2</sub> with *Rhodobacter sphaeroides*

#### Planktonic cell attach to electrode surface

#### Coulomb Recovery into H<sub>2</sub>



Shuwei Li and Jung Rae Kim. 2021. Bioresource Technology. 320:124333

añe



ISSN 1385-8947

#### **Polyhydroxybutyrate (PHB) Production**

PHB

Shuwei Li & Jung Rae Kim et al. Electron uptake from solid electrodes promotes the more efficient conversion of CO2 to polyhydroxybutyrate by using *Rhodobacter sphaeroides*. Chemical Engineering Journal 469 (2023) 143785

PHB









#### Electrode-driven regulation of CO<sub>2</sub> conversion pathway

Shuwei Li and Jung Rae Kim. 2023. Chemical Engineering Journal. 469:143785



Shuwei Li and Jung Rae Kim. 2023. Chemical Engineering Journal. 469:143785

#### Attachement of cell on electrode surface



# *R. Sphaeroides* stick to the various carbon surface in MES

- Carbon rod
- ITO
- Titanium mesh
- Glassy carbon

#### FE-SEM of electrode-associated cell



*R. Sphaeroides* may uptake electron directly from the cathode electrode through Fla2 flagella sysem

Shuwei Li and Jung Rae Kim. 2023. Chemical Engineering Journal. 469:143785



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#### Pathway of PHB synthesis from CO<sub>2</sub>



#### Relative gene expression profile





## Scaled-up MEC Reactor for H<sub>2</sub> Production

#### Two-chamber MEC (8L)









#### Single-chamber MEC (5L)





## **Upgrading AD Biogas with MEC**





Minsoo Kim and Jung Rae Kim. 2022. Chemical Engineering Journal 446:137079

## **Upgrading AD Biogas with MEC**



#### Conversion of CO<sub>2</sub> to CH<sub>4</sub> in scale-up MEC reactor



#### 100 1.6 80 1.5 Total gas volume (L) 60 1.4 40 20 1.3 0 Initial 1 2 3 5 4 Time (day)

#### Cost estimation of MES CH<sub>4</sub> production

\* Below 300 kWh industrial, ~ 6 cent/ kWh

Reactor Volume	Applied Potential (V vs Ag/AgCl)	Temp.	CH₄ Production rate (L CH₄/m² cat/hour)	CH₄ Production per energy (L CH₄/kWh)	Cost of CH <sub>4</sub> production (USD/Nm <sup>3</sup> CH <sub>4</sub> )
250 + 250 mL	-1.0V	30	2.9	259	0.24 USD
3L + 3L	-1.0V	40	2.29	14.9	4.13 USD

Gas content (%)

Calculated at standard temperature and pressure (STP) Calculated with the immersed cathode surface area.

반응기에 인가되는 전압대비, 메탄생성량을 이용하여 계산하였음. (설치비, 운영비, 약품비, 인건비, Utility관리비 제외)

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## CO<sub>2</sub> to CH<sub>4</sub> with MES



#### 전극상 메탄생산 미생물의 고농도 배양



- CH<sub>4</sub> 전환 반응기에서 각각 메탄생성균이 부착된 전극과 배양액을 분리하여 Serum bottle에 배양함.
- Serum bottle 조건의 경우 전자(Electron) 공급을 위해 Potentiostat 대신  $H_2$ 를 공급하였음.
- Methanogen attached electrode만 넣은 Serum bottle에서 CO<sub>2</sub>와 H<sub>2</sub>의 소모를 확인하였으며, 이는 <u>메탄</u>
  <u>생성에서 전극상 부착된 군집이 주요한 역할을 하고 있음</u>을 나타냄.

## **Effect of Hydrodynamic Shear Stress**



### **Microbe-Electrode Interaction**



## 1. 미생물-전극 반응을 어떻게 분석할 것인가?

## 2. 생물전기화학시스템으로 무엇을 생산할 것인가?

### 3. 생물전기화학시스템의 실용화 전략은?

→ 미생물-전극 분석 및 제어

- → Carbon capture & utilization (CCU)
- $\rightarrow$  Electro-fermentation
- $\rightarrow$  Power to Product (P2P)

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# 연구진 구성





### **Biofunctionalization of Electrode Surface by Live Cell**



May et al. 2016. Curr Opin Biotechnology

### PNU Bioenergy & Bioprocess Engineering Lab (BBE Lab)

http://bioenergy.pusan.ac.kr/

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